CLAIMS

1	1. A double-stranded conducting polymer, said polymer selected from the group
2	consisting of Polyaniline:Poly(vinylphosphate) double-stranded complex,
3	Polyaniline:Poly(acrylic acid-co-vinylphosphate) complex, Polyaniline:Poly(mathacrylic acid-
4	co-vinylphosphate) complex, Polypyrrole:Poly(vinylphosphate) double-stranded complex,
5	Polypyrrole:Poly(acrylic acid-co-vinylphosphate) complex, Polypyrrole:Poly(vinylmathacrylic
6	acid-co-vinylphosphate complex, Polyaniline:Poly(methylacrylate-co-vinylphosphate) complex,
7	Polypyrrole:Poly(methylacrylate-co-vinylphosphate) complex, Polyaniline:Poly(butylacrylate-
8	co-vinylphosphate) complex, and Polypyrrole:Poly(butylacrylate-co-vinylphosphate) complex.

- 1 2. The double-stranded conducting polymer of claim 1, wherein a first strand is a reversible electron donor or acceptor.
- 1 3. The double-stranded conducting polymer of claim 1, wherein a second strand 2 includes the integration of appropriate ligands.
- 1 4. The double-stranded conducting polymer of claim 2, wherein the ligand is a carboxylic or phosphate functional group.
- 5. A composition including a conducting polymer, said composition comprising:
 polyaniline or polypyrrole, Poly(vinyl butyral), molybdenum oxide or cerium oxide magnesium
- 3 silicate, carbon black or lamp black, n-butyl alcohol, isopropyl alcohol, and water.

- 1 6. The composition of claim 5, further comprising phosphoric acid, water, and 2 isopropyl alcohol.
- 7. A composition including a conducting polymer to treat metal surfaces to provide
 a stable interface for adhesive binding or coating.
- 1 8. A formulation for surface treatment reagents which includes a double-stranded 2 conductive polymer.
- 1 9. The use of water-borne double-stranded conducting polymers for as a surface 2 conversion or surface treatment agent for metal surfaces, as a early-warning indicator for metal 3 corrosion, as a component for a wash primer for aluminum alloys, magnesium alloys, steel and 4 other non-noble metals, as a surface modification coating on non-metallic surfaces to catalyze 5 deposition of decorative and functional top coatings, as an additive to improve the performance 6 of adhesive bonding of metals, or for others that are logical extensions of the above application.